

Finally, the acetyl groups was located on the glucosyl C-2 hydroxy group by a double resonance experiment irradiation of the glucosyl H-1 converted the double doublet at  $\delta$  5.68 into a simple doublet

### EXPERIMENTAL

The plant material was provided by Mr M F De Carvalho, Scientific Investigation Institute, Agronomic Section, Lourenço Marques (Mozambique). A herbarium specimen is available at the Department of Pharmacognosy, Università della Beffa, Milan, Italy.

MS were corrected. NMR were recorded on a Varian XL-100 instrument. MS were obtained on a Varian CH7 spectrometer and IR on a Perkin-Elmer model 157G instrument. UV was measured on a DBG-T Beckmann spectrophotometer.

**Isolation of flavonoids from *B. zanguebarica*.** The dry leaves were extracted at 30° with 90% aq. MeOH. After concentration, extraction with  $\text{CHCl}_3$  and column chromatography [eluent:  $\text{AcOEt-EtOH-H}_2\text{O}$  (100:13.5:10), silica gel] afforded *sequoiaflavone* and *vitexin 2''-O-acetyl 7-O-methyl ether* (1). Extraction with *n*-BuOH yielded, after column chromatography [silica gel, eluent:  $\text{CHCl}_3\text{-MeOH-H}_2\text{O}$  (13:7:2), lower phase], *vitexin* and *isoorientin*.

***Vitexin 2''-O-acetyl 7-O-methyl ether*.** The flavonoid (1) showed the following properties: m.p. 180° ( $\text{AcOEt}$ ),  $[\alpha]_D^{25} -63^\circ$  (c 0.5, pyr),  $\lambda_{\text{max}}$  (MeOH) 266, 298 sh, 333 nm;  $\lambda_{\text{max}}$  (MeONa) 253, 268, 298 sh, 388 nm;  $\lambda_{\text{max}}$  ( $\text{AlCl}_3$ ) 274, 302, 340, 387 nm;  $\lambda_{\text{max}}$  ( $\text{AlCl}_3\text{-HCl}$ ) 274, 300, 337, 383 nm;  $\lambda_{\text{max}}$  ( $\text{AcONa}$ ) 258 sh, 268, 299 sh, 391 nm;  $\lambda_{\text{max}}$  ( $\text{AcONa-H}_3\text{BO}_3$ ) 267, 298 sh, 337 nm; MS  $m/e$  488 ( $\text{M}^+$ ), 469, 445, 428, 427, 413, 396, 326, 313, 297, 255, 179, 118, 121; IR (KBr) 3400, 1745, 1655  $\text{cm}^{-1}$ .

***Vitexin 7-O-methyl ether*.** The HCl-hydrolyzed product from (1) showed the following properties: m.p. 200° ( $\text{H}_2\text{O}$ ),  $\text{M}^+ = 466$ , NMR signals after silylation ( $\text{CCl}_4$ ,  $\delta$ ) 7.87 (*d*, *J* 10 Hz, H-2' and H-6'), 6.84 (*d*, *J* 10 Hz, H-3' and H-5'), 6.34 (*s*, H-3), 6.24 (*s*, H-6), 4.89 (*d*, *J* 10 Hz, H-1'), 3.83 (*s*, MeO-) and six protons between 4.4 and 3.1; IR (KBr) 3400, 1655  $\text{cm}^{-1}$ .

**Oxidation of (1) with  $\text{FeCl}_3$ .** was carried out in aq. soln. by heating in an oil-bath at 125° for 6 hr. Filtration through a column of silica gel using water as eluent and concentration yielded a syrup which on PC showed identity with glucose.

**Acetylation of both (1) and the HCl-hydrolyzed product (3).** m.p. 177° (*iso*- $\text{Pr}_2\text{O}$ ). NMR acetate signals ( $\text{CDCl}_3$ ,  $\delta$ ) at 2.40 (C-5), 2.30 (C-4), 2.05 (C-4'), 1.95 (C-3'), 1.87 (C-6'), 1.70 (C-2').

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## FUROCHROMONES OF *ERANTHIS PINNATIFIDA*

HIROSHI WADA, MITSUNORI GAINO and SEIICHI SAITO

Organic Chemistry Research Laboratory Tanabe Seiyaku Co. Ltd. Kawagishi 2-2-50 Toda Saitama Japan

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**Key Word Index**—*Eranthis pinnatifida*, Ranunculaceae, khellol, norkhellol, cimifugin, new furochromones.

In earlier work on the chromone constituents of *Eranthis*, khellol glucoside was only found in *E. hyemalis*.<sup>1</sup> In this communication we wish to report five chromones from *E. pinnatifida* Maxim. all of which have 7-hydroxymethyl groups.

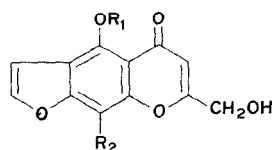
The MeOH extract of the leaves and stems collected at the flowering season afforded, after chromatographic separation, five chromones, khellol (1),<sup>2</sup> norkhellol (2)<sup>3</sup> and three

<sup>1</sup> EGGER, K. (1961) *Z. Naturforsch.* **16b**, 697.

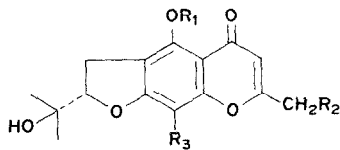
<sup>2</sup> SPATH, E. and GRUBER, W. (1941) *Chem. Ber.* **74**, 1549.

<sup>3</sup> SCHONBERG, A. and AZIZ, G. (1953) *J. Am. Chem. Soc.* **75**, 3265.

new chromones, cimifugin (3) m p 128–130°, norcimifugin (4) m p 191° and norammol (5) m p 242°



- (1)  $R_1 = \text{Me}$ ,  $R_2 = \text{H}$   
 (2)  $R_1 = \text{H}$ ,  $R_2 = \text{H}$   
 (5)  $R_1 = \text{H}$ ,  $R_2 = \text{OMe}$



- (3)  $R_1 = \text{Me}$ ,  $R_2 = \text{OH}$ ,  $R_3 = \text{H}$   
 (4)  $R_1 = \text{H}$ ,  $R_2 = \text{OH}$ ,  $R_3 = \text{H}$   
 (7)  $R_1 = \text{OMe}$ ,  $R_2 = \text{Cl}$ ,  $R_3 = \text{H}$   
 (8)  $R_1 = \text{H}$ ,  $R_2 = \text{H}$ ,  $R_3 = \text{H}$   
 (9)  $R_1 = \text{H}$ ,  $R_2 = \text{H}$ ,  $R_3 = \text{I}$

The first crystalline compound (3)  $\text{C}_{16}\text{H}_{18}\text{O}_6$ , m p 128–130° on recrystallization from  $\text{C}_6\text{H}_6\text{--Me}_2\text{CO}$ , whose hydrated sample m p 107° was also prepared from aq MeOH, showed  $[\alpha]_D + 78^\circ$ , characteristic UV [ $\lambda_{\text{max}}^{\text{MeOH}}$  229, 250 (inf), 293 nm] and IR ( $\nu_{\text{max}}^{\text{KBr}}$  3380, 1660, 1610  $\text{cm}^{-1}$ ). The NMR spectrum<sup>4</sup> indicated the presence of two tertiary methyl groups (1.36, 6H, s), a methylene group (3.45, 2H, d, J 9), a methoxyl group (4.06, 3H, s), a hydroxymethyl group (a methylene group 4.55, 2H, d, J 6.2, a hydroxyl group 5.91, 1H, t, J 6.2), a tertiary hydroxyl group (4.93, 1H, s), a methine proton (4.95, 1H, t, J 9) and two aromatic protons (6.3, 1H, s, 6.83, 1H, s). The structure of 3, which was deduced from the spectroscopic data, was further established by transformation to visammol (6). The treatment of 3 with mesyl chloride in pyridine afforded a chloride (7) which was reduced and demethylated with NaI in refluxing AcOH to give a deoxy compound (8)  $\text{C}_{15}\text{H}_{16}\text{O}_5$ , accompanied with an iodide (9)<sup>5</sup>  $\text{C}_{15}\text{H}_{15}\text{IO}_5$ , m p 147°. The deoxy compound 8 was identical with visammol (6)<sup>6</sup> by m m p and IR, MS spectral comparisons. Therefore the structure of 3 was concluded to be 2,3-dihydro-2-isopropoxy-4-methoxy-7-hydroxymethyl-furochromone. When our work was in progress the structure of 3 was given by Takemoto for cimifugin<sup>7</sup> m p 107° which was isolated from *Cimicifuga simplex* Wormsk. The hydrated sample of 3, m p 107° was found to be identical with an authentic sample of cimifugin by m m p and IR, MS spectral comparisons.

The second new compound (4)  $\text{C}_{15}\text{H}_{16}\text{O}_6$ , m p 191°,  $[\alpha]_D + 49^\circ$  showed positive  $\text{FeCl}_3$  color test and characteristic UV [ $\lambda_{\text{max}}^{\text{MeOH}}$  231, 251, 257 (inf), 299 nm]. The NMR spectrum of 4 was almost identical with that of cimifugin (3), but the signal of methoxyl group (4.06) in 3 was replaced by that of a phenolic hydroxyl group (12.76) in 4. In addition to this, the fact that methylation of 4 gave cimifugin (3) proved 4 to be 4-desmethyl cimifugin and therefore it was designated as norcimifugin.

The third new compound (5)  $\text{C}_{13}\text{H}_{10}\text{O}_6$ , m p 242° (dec) showed positive  $\text{FeCl}_3$  color test and very similar UV absorption to those of 4-desmethyl khellin<sup>6</sup>. The NMR spectrum indicated the presence of a methoxyl group (3.77, 3H, s), a hydroxymethyl group (a methylene group 4.35, 2H, s, a hydroxyl group 5.60), an olefinic proton on the  $\gamma$ -pyrone ring

<sup>4</sup> Chemical shifts are given in ppm from the internal standard TMS in  $\text{DMSO}-d_6$  at 100 MHz. Coupling constants in Hz.

<sup>5</sup> Since the C-6 proton signal (6.05 ppm, d, J 1), coupled with C-7 methyl group (2.37 ppm, d, J 1) was observed in the NMR spectrum while the C-8 aromatic proton signal was absent, iodine located at C-8 in the dihydro-furochromone skeleton of 9.

<sup>6</sup> BENČE, W., EISENBEISS, J. and SCHMID, H. (1956) *Helv. Chim. Acta* **39**, 923.

<sup>7</sup> KONDO, Y. and TAKEMOTO, T. (1972) *Chem. Pharm. Bull.* **20**, 1940.

(6.05, 1H, s), two protons on the furan ring (7.05, 1H, *d*, *J* 2, 7.95, 1H, *d*, *J* 2) and a chelated hydroxyl group (10.10). In comparison with the NMR spectrum of khellol (**1**), the absence of a proton signal on the benzene ring at 7.20–7.33 ppm region in the spectrum of **5** suggested that oxygen functions located at both positions C-4 and C-8 in the furochromone structure of **5**. From these findings **5** was deduced to be 4-desmethyl ammiol and hence it was named as norammiol.

#### EXPERIMENTAL

**Extraction and isolation procedure.** Fresh leaves and stems (1.85 kg) were extracted with MeOH. The filtered MeOH extract was concentrated to a small vol., washed repeatedly with *n*-hexane, and taken up in EtOAc. The EtOAc solution was shaken with H<sub>2</sub>O, dried and concentrated to a syrup which was dissolved in CHCl<sub>3</sub> and poured on a column of silica gel (300 g). Elution with CHCl<sub>3</sub> and increasing amount of Me<sub>2</sub>CO in CHCl<sub>3</sub> afforded the following fractions each of which was examined by TLC [silica gel G, CHCl<sub>3</sub>–Me<sub>2</sub>CO, 1:1]. (a) Norkhellol (**2**), eluted with CHCl<sub>3</sub>–Me<sub>2</sub>CO (10:1), a colorless needles (200 mg) from MeOH–THF, m.p. 200–202°. IR  $\nu_{\text{max}}^{\text{KBr}}$  3360, 1660, 1625, 1590 cm<sup>-1</sup>. NMR ( $\delta$  in DMSO-*d*<sub>6</sub>) 4.40 (2H, *d*, *J* 6), 5.80 (1H, *t*, *J* 6, OH), 6.28 (1H, *s*), 7.03 (1H, *d*, *J* 3), 7.20 (1H, *s*), 7.95 (1H, *d*, *J* 3), 13.35 (1H, *s*, OH). UV  $\lambda_{\text{max}}^{\text{MeOH}}$  nm ( $\epsilon$ ) 245 (37,700), 253 (43,500), 262 (inf 29,600), 285 (inf 6,400). MS (*m/e*) 232 (*M*<sup>+</sup>). (b) Norcimifugin (**4**), eluted with CHCl<sub>3</sub>–Me<sub>2</sub>CO (8:1), m.p. 191–192° (930 mg) from MeOH. IR  $\nu_{\text{max}}^{\text{KBr}}$  3250 (br.), 1665, 1625, 1595 cm<sup>-1</sup>. MS (*m/e*) 292 (*M*<sup>+</sup>) (Found C, 61.42, H, 5.58. C<sub>15</sub>H<sub>16</sub>O<sub>6</sub> requires C, 61.70, H, 5.52%). (c) Khellol (**1**), later fractions eluted with CHCl<sub>3</sub>–Me<sub>2</sub>CO (8:1), m.p. 175–176° (2.61 g) from MeOH. (d) Norammiol (**5**), eluted with CHCl<sub>3</sub>–Me<sub>2</sub>CO (5:1), needles (130 mg) from MeOH–THF, m.p. 240–242° (dec.). IR  $\nu_{\text{max}}^{\text{KBr}}$  3400–3100 (br.), 1655, 1605 cm<sup>-1</sup>. MS (*m/e*) 262 (*M*<sup>+</sup>) (Found C, 59.79, H, 4.01. C<sub>13</sub>H<sub>10</sub>O<sub>6</sub> requires C, 59.79, H, 3.85%). (e) Cimifugin (**3**), eluted with CHCl<sub>3</sub>–Me<sub>2</sub>CO (4:1), m.p. 128–130° (594 mg) from C<sub>6</sub>H<sub>6</sub>–Me<sub>2</sub>CO which on recrystallization from aq MeOH gave needles m.p. and m.m.p. 107°, identical IR and MS spectra with an authentic sample of cimifugin.

**Reduction of 3.** To a solution of **3** in pyridine (153 mg/3 ml) mesyl chloride (63 mg) was added at 5°. The mixture was kept for 12 hr at the temp. and poured onto ice. The aqueous phase was extracted repeatedly with CH<sub>2</sub>Cl<sub>2</sub>. The combined CH<sub>2</sub>Cl<sub>2</sub> extracts were washed with 0.2 N HCl, aq NaHCO<sub>3</sub>, saturated aq NaCl, dried (Na<sub>2</sub>SO<sub>4</sub>) and concentrated to give a resinous product which was purified by preparative TLC (silica gel, CHCl<sub>3</sub>–MeOH, 100:1). An amorphous chloride (70 mg), MS (*m/e*) 326, 324 (*M*<sup>+</sup>) was obtained. A soln. of the chloride (110 mg) and NaI (400 mg) in AcOH (5 ml) was refluxed for 16 hr. The reaction mixture was poured into H<sub>2</sub>O and extracted with CH<sub>2</sub>Cl<sub>2</sub>. The CH<sub>2</sub>Cl<sub>2</sub> extract, after washing with aq NaHCO<sub>3</sub>, 3% aq Na<sub>2</sub>S<sub>2</sub>O<sub>3</sub>, aq NaCl and dried with Na<sub>2</sub>SO<sub>4</sub>, was concentrated to give a glass which was separated by preparative TLC (silica gel, C<sub>6</sub>H<sub>6</sub>–EtOAc, 1:1). The more mobile compound was an iodide (**9**) m.p. 147° (18 mg) from Et<sub>2</sub>O. IR  $\nu_{\text{max}}^{\text{CHCl}_3}$  3580, 3400, 1660, 1625, 1590 cm<sup>-1</sup>. MS (*m/e*) 401 (*M*<sup>+</sup>). NMR ( $\delta$  in CDCl<sub>3</sub>) 1.21 (3H, *s*), 1.36 (3H, *s*), 1.90 (1H, *br s*, OH), 2.37 (3H, *d*, *J* 1), 3.24 (2H, *d*, *J* 8), 4.80 (1H, *t*, *J* 8), 6.05 (1H, *d*, *J* 1), 12.91 (1H, *s*, OH). UV  $\lambda_{\text{max}}^{\text{MeOH}}$  nm ( $\epsilon$ ) 216 (31,000), 233 (inf 21,700), 262 (20,100) (Found C, 44.59, H, 3.74, I, 31.63. C<sub>15</sub>H<sub>15</sub>IO<sub>5</sub> requires C, 44.93, H 3.77, I, 31.65%). The less mobile was a deoxy compound (**8**) m.p. 157–159° (61 mg) from EtOAc which was identical with an authentic sample of visammitol by m.p. and IR, MS comparisons.

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